

Al-Co-Fe (Aluminum-Cobalt-Iron)

V. Raghavan

The phase relationships in this system were reviewed by [1988Ray], who presented two partial liquidus surfaces, an isothermal section at 800 °C for Co-rich alloys and isothermal sections at 640 and 600 °C for alloys near the Al corner. A full isothermal section at 650 °C from [1999Koz] and a partial section at 1127 °C were reviewed by [2002Rag]. [2004Kam] determined the phase relationships in Co-rich alloys between 1200 and 900 °C in the region of magnetic and order-disorder transitions, briefly reviewed by [2005Rag]. Recently, [2004Koz] and [2006Koz] used the liquid-quenching and composition-gradient methods to determine the metastable $B2/(B2 + A2)$ phase boundary at 650 °C in Co-rich alloys.

Binary Systems

In the Al-Co phase diagram [Massalski2], four intermediate phases, Al_9Co_2 , $Al_{13}Co_4$, Al_3Co , and Al_5Co_2 , occur in the Al-rich region. With increasing Co, $AlCo$ ($B2$, CsCl-type cubic) occurs over a wide temperature and composition range (48-78 at.% Co). In the Fe-Al phase diagram [Massalski2], the solid solution based on the face-centered cubic (fcc) Fe is restricted by a γ loop. The solution based on the body-centered cubic (bcc) Fe exists in the disordered ($A2$) and the ordered ($B2$ and $D0_3$) forms. With increasing Al, four intermediate phases occur: ϵ , $FeAl_2$, Fe_2Al_5 and $FeAl_3$. In the Co-Fe system [Massalski2], a continuous fcc solid solution γ between fcc Fe and Co is stable over a wide temperature range. The fcc \rightarrow bcc transformation temperature in Fe is initially raised by the addition of Co, reaching a maximum at 980 °C at 45 at.% Co. At 730 °C, the bcc phase of equiatomic composition orders to a $B2$ structure via a second-order transition.

Ternary Isothermal Section

With starting metals of 99.99% Al, 99.9% Co, and 99.98% Fe, [2006Koz] arc-melted under Ar atm five Co-rich ternary alloys containing 9.58-37 at.% Al and 6.11-7.26 at.% Fe. Liquid-quenching was one of the methods adopted to obtain the metastable condition. The alloys were remelted in a quartz tube and the melt was poured through a nozzle on to a steel roller rotating at high speed to produce ribbons. In the second method, a composition gradient was obtained in a sample through an appropriate heat treatment procedure. The phase equilibria were studied with x-ray powder diffraction and transmission electron microscopy. The compositions were measured with energy dispersive spectroscopy. The metastable isothermal

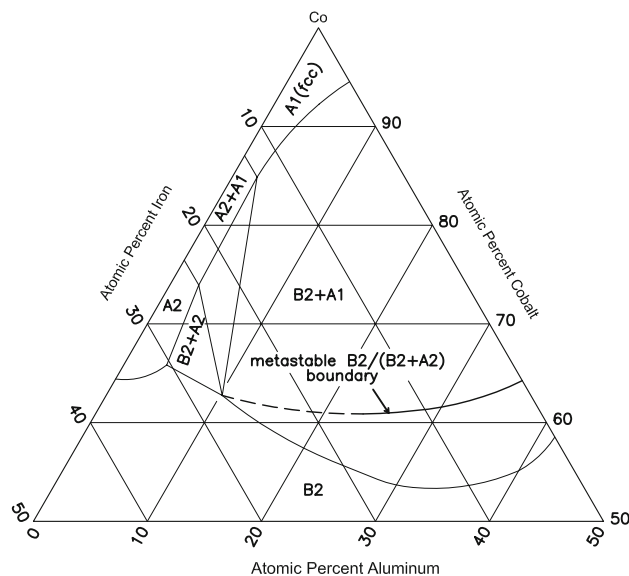


Fig. 1 Al-Co-Fe metastable $B2/(B2 + A2)$ phase boundary in the Co-rich region at 650 °C [2006Koz]

section near the Co corner at 650 °C constructed by [2006Koz] is shown in Fig. 1. The thin lines correspond to stable equilibrium [1999Koz] and the thick line represents the metastable $B2/(B2 + A2)$ phase boundary. The extent of the $B2$ field is increased under metastable conditions.

References

- 1988Ray:** G.V. Raynor and V.G. Rivlin, Al-Co-Fe, *Phase Equilibria in Iron Ternary Alloys*, Institute of Metals, London, 1988, p 71-81
- 1999Koz:** T. Kozakai and T. Miyazaki, Phase Equilibria in the Fe-Al-Co Ternary System at 923 K, *Z. Metallkd.*, 1999, **90**(4), p 261-266
- 2002Rag:** V. Raghavan, Al-Co-Fe (Aluminum-Cobalt-Iron), *J. Phase Equilib.*, 2002, **23**(5), p 434-436
- 2004Kam:** N. Kamiya, T. Sakai, R. Kainuma, I. Ohnuma, and K. Ishida, Phase Separation of BCC Phase in the Co-Rich Portion of Co-Fe-Al System, *Intermetallics*, 2004, **12**, p 417-423
- 2004Koz:** T. Kozakai, T. Shikama, T. Koyama, and M. Doi, Metastable Two-Phase Field ($A2 + B2$) in Co-Al-Fe and Co-Al Alloy Systems, *Mater. Sci. Forum*, 2004, **449-452**, p 61-64
- 2005Rag:** V. Raghavan, Al-Co-Fe (Aluminum-Cobalt-Iron), *J. Phase Equilib. Diffus.*, 2005, **26**(1), p 57-58
- 2006Koz:** T. Kozakai, T. Koyama, and M. Doi, Phase Decomposition and Precipitation of Metastable $A2$ Phase in $B2$ Ordered Co-Al-Fe Alloys, *Z. Metallkd.*, 2006, **97**(3), p 266-272